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Regiospecific *ortho*-Lithiation of Phthalocyanines: a Route for One-step Synthesis of Highly Soluble Trimethylsilyl Substituted Phthalocyanines

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In situ trapping of lithiated phthalocyanines by chlorotrimethylsilane produces highly soluble phthalocyanines; the reaction is regiospecific and yields non-peripherally (1, 4, 8, 11, 15, 18, 22, and 25 positions) substituted phthalocyanines, $H_2Pc(SiMe_3)_x$ where x = 4, 3, and 2.

Low solubility in common organic solvents is a characteristic of phthalocyanines (1) that has hampered the development of their solution chemistry. Most of the effort to solubilize phthalocyanines has been directed towards synthesis of peripherally (2, 3, 9, 10, 16, 17, 23, and 24 positions) substituted derivatives.¹⁻⁵ A recent paper describes octa-alkylphthalocyanines with substitution at non-peripheral (1, 4, 8, 11, 15, 18, 22, and 25) positions.⁶ All of these phthalocyanines were prepared by condensation of precursors, *e.g.*, substituted phthalonitriles and phthalimides, preparations of which are often laborious. We report here a one-step synthesis that converts preformed phthalocyanines into highly soluble non-peripherally substituted phthalocyanines.

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Figure 1. ¹H N.m.r. spectra (300 MHz) of (A) Li₂Pc and (B) deuteriated Li₂Pc in [²H₆]-acetone



Addition of lithium 2,2,6,6-tetramethylpiperidide (Li-TMP; 7.8 mmol) to a solution of 29H, 31H-phthalocyanine (H₂Pc, 1.0 mmol) and chlorotrimethylsilane (8.7 mmol) in tetrahydrofuran at -78 °C, followed by slow warming to ambient temperature, and acidification, produced a mixture of trimethylsilyl-substituted phthalocyanines, H₂Pc(SiMe₃)_x. The product mixture, separated *via* column chromatography, gave isolated yields of 13, 18, and 3% for x = 4, 3, and 2, respectively.* No H₂Pc(SiMe₃)_x with x > 4 was obtained even for a reaction with a ratio of Li-TMP/H₂Pc = 32; under these conditions, the yield of x = 4 product was increased. In this reaction, the sterically hindered Li-TMP, instead of attacking Me₃SiCl, *ortho*-lithiates H₂Pc, which is then trapped by Me₃SiCl to afford the products.⁷ The product mixture is very soluble in nonpolar organic solvents. For example, the solubility in CH_2Cl_2 is 0.07M at 25 °C.

The ¹H n.m.r. spectra of the aromatic protons of $Li_2Pc(SiMe_3)_4$ show that there are only half as many nonperipheral protons ($\delta = 9.34$ —9.41 p.p.m.) as the peripheral protons ($\delta = 7.98$ –8.05 p.p.m.), indicating that the lithiation and, therefore, the silvlation are directed very selectively, if not entirely, to the non-peripheral positions. Such heteroatom directed ortho-lithiations are well known in aromatic and heteroaromatic systems.⁸ In support of this conclusion, a Li-TMP catalyzed H–D exchange reaction at -23 °C between Li₂Pc and [²H]-2,2,6,6-tetramethylpiperidine, leads to selective replacement of the non-peripheral hydrogens. As expected, the ¹H n.m.r. spectra of the deuteriated Li₂Pc (Figure 1) shows a singlet at δ 8.02 p.p.m. for the peripheral protons, because their couplings to the non-peripheral protons vanish; the multiplet at δ 9.37 p.p.m. for the residual non-peripheral protons is relatively weak. The ²H n.m.r. spectra of the deuteriated Li₂Pc show a strong singlet at δ 9.38 p.p.m. for the non-peripheral deuteriums while the expected singlet at δ 8.02 p.p.m. for the peripheral



deuteriums is barely discernible. At higher reaction temperatures, the selectivity is poorer. The fact that all eight nonperipheral protons of Li_2Pc are replaced in the H–D exchange experiment but only a maximum of four trimethylsilyl groups may be put on these positions suggests that only one trimethylsilyl group may occupy each of the pockets between two adjacent isoindole moieties, presumably due to the bulkiness of the group. With this constraint, there are four isomers for H₂Pc(SiMe₃)₄ which have a statistical distribution

^{*} All these products were isolated as isomeric mixtures from which some pure isomers may be isolated. The analytical data for $H_2Pc(SiMe_3)_x$ fractions are presented below. (1) For x = 4, Found: C, 65.8; H, 6.35; N, 14.05; Si, 13.4. Calc. for: C, 65.8; H, 6.25; N, 13.95, Si, 14.00%. (2) For x = 3, Found: C, 67.25; H, 5.9; N, 15.6; Si, 11.51. Calc. for: C, 67.35; H, 5.8; N, 15.35; Si 11.55% (3) For x = 2, Found: C, 69.2; H, 5.2; N, 16.95; Si, 8.65. Calc. for: C, 69.25; H, 5.20; N, 17.0; Si, 8.55%.



Figure 2. ²⁹Si{¹H} N.m.r. spectrum (60 MHz; TMS at $\delta = 0$) of H₂Pc(SiMe₃)₄ in CDCl₃

of 1:1:2:4. An isomeric mixture of this distribution would have eight ²⁹Si resonances of equal intensity for the eight magnetically distinct SiMe₃ groups. Consistent with this, the ²⁹Si n.m.r. spectrum of H₂Pc(SiMe₃)₄ (Figure 2) shows 8 resonances at $-2.515 \ge \delta \ge -2.533$ p.p.m. (in reference to TMS in CDCl₃ at $\delta = 0$). The distribution of isomers appears to be close to statistical (within a factor of 3). For the purpose of comparison, one of the four isomers, 1,8,15,22-tetrakis-(trimethylsilyl)phthalocyanine, has been prepared by a more laborious synthetic route.* This isomer gives a single 29 Si resonance at $\delta - 2.366$ p.p.m.

* Synthesis of this compound will be reported separately.

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